

Major oxide geochemistry of Okeigbo stream sediments in Akungba-Akoko, southwestern Nigeria: implications for chemical classification and provenance

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Abstract

In this paper, major oxide concentrations in Okeigbo stream sediments were used to chemically classify them and to determine their provenance. The study area lies in Akungba-Akoko area of Ondo State, southwestern Nigeria. Seventeen samples were collected and analyzed using Energy Dispersive X-ray Fluorescence Spectrometer. The major oxides analyzed in the samples are SiO₂, Al₂O₃, Fe₂O₃, TiO₂, K₂O, CaO, MnO and P₂O₅. The average concentration values of SiO₂, Al₂O₃ and Fe₂O₃ are 56.50 wt. %, 13.50 wt. % and 10.70 wt. %, respectively. On the other hand, the average concentration values of K₂O and CaO are 6.30 wt. % and 4.99 wt. %, respectively, while TiO₂, MnO and P₂O₅ have values < 1.00 wt. %. Okeigbo stream sediments are classified using the plot of log₁₀(SiO₂/Al₂O₃) versus log₁₀(Fe₂O₃/K₂O) to be 89.0% greywackes, 10.0% litharenites and 10.0% arkose. The average value of 4.28 for SiO₂/Al₂O₃ indicates that the sediments originated from majorly acid igneous source rock, while the average value of 1.73 for Fe₂O₃/K₂O is indicative of mineralogically stable sediments.

Keywords: Akungba-Akoko; Geochemistry; Major Oxides; Okeigbo; Stream Sediments.

1. Introduction

Stream sediments have been used in different parts of the world to study the geology, environmental geochemistry, mineral exploration, provenance of the sediments of an area [1-7]. Stream sediments serve as ultimate sinks for elements in aquatic setting. Stream sediments are employed almost exclusively for reconnaissance geochemical studies in drainage basins [8].

The study area is located in Okeigbo area of Akungba-Akoko, which is a town in Ondo State southwestern Nigeria (Fig. 1). The aim of this paper is to present the use of major oxide concentrations in stream sediments to chemically classify and decipher their provenance. The study area is situated within the tropical rainforest of southwestern Nigeria [9].

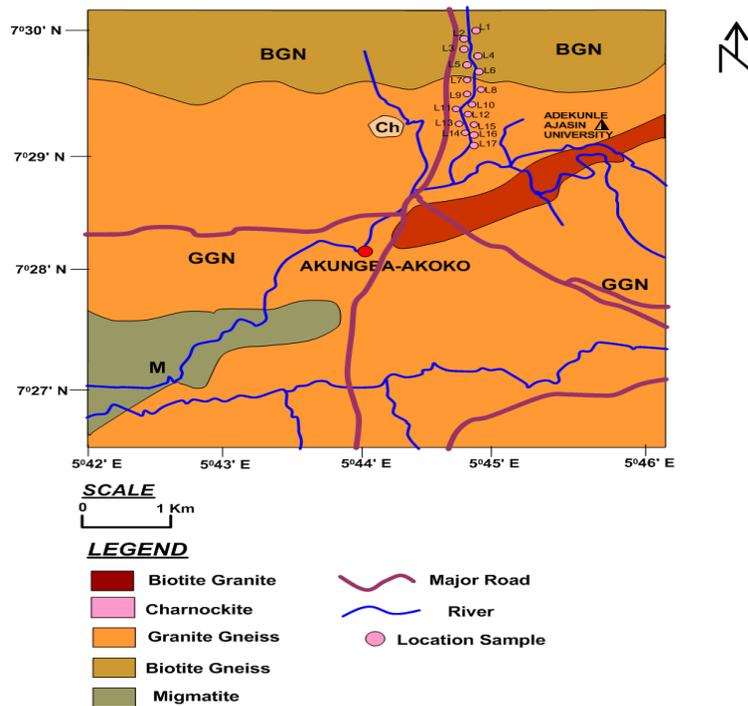


Fig. 1: Geological Map of Akungba-Akoko Showing Samples Locations (After Ogunyele et al. [10]).

1.1. Regional geology

The study area lies within the crystalline Basement Complex rocks of southwestern Nigeria [10-13]. The Nigerian Basement Complex is subdivided into six groups, namely: (1) migmatite gneiss quartzite complex, (2) meta-sedimentary schist belts, (3) older granite, (4) charnockitic or charnockitic gabbroic and dioritic rocks, (5) metamorphosed and unmetamorphosed calc-alkaline volcanic and hyperbyssal rocks and lastly, (6) metamorphosed dolerite dykes, syenite dykes [11].

The migmatite gneiss quartzite complex occupy vast portion in Nigeria [11]. The constituent rocks of this complex are heterogeneous and made up mainly of migmatites, granitic gneisses, basic schists, gneisses and relict meta-sedimentary calcareous quartzitic and granulitic rocks. The rocks of this complex were further classed into three main groups [14] such as (a) migmatitic unit consisting largely of paleosomatic, banded para-gneisses, banded ironstones, ferruginous quartzite that have been mixed with quartzites, granites and aplites; (b) gneissic unit made of hornblende granites and augen gneisses, and (c) mafic unit made up of calcareous and biotite-plagioclase schists.

The second group, meta-sedimentary schist belts have lithologic similarities to schist belts from other parts of the world and they are host rocks to very importance economic mineral deposits [11].

On the other hand, the older granites, which are Pan-African in age [10-13], constitute around 40-50% of the Nigerian Basement Complex and >20% in southwestern Nigeria. Older granites refer to rocks such as tonalites, true granites, granodiorites and syenites [11].

2. Materials and methods

Seventeen samples were taken from Okeigbo stream sediments around Akungba-Akoko covering latitudes $N07^{\circ} 29' 10''$ and $N07^{\circ} 29' 39''$ and longitudes $E05^{\circ} 44' 13''$ and $E05^{\circ} 44' 42''$ at mean elevation of 416m above the sea level. The samples were packed into labeled Kraft envelopes covered with polythene bags and transported to the laboratory for treatment and 60 μm size grains of the samples were analyzed for major oxide elements using X-ray fluorescence (XRF) spectrometer at NASENI EDMI Center of Excellence in Nanotechnology Laboratory in Akure, Nigeria.

Each sample was pulverized and then dried in an oven at 110°C for 24 hours to remove its moisture content and thereafter converted to beads for major elemental analysis expressed in oxide weight percent. Furthermore, 5 grams of each dried sample was weighed in the silica crucible and ignited in the furnace at $1,000^{\circ}\text{C}$ for 2 to 3 hours to calculate the impurities in each sample powder. Thereafter, each sample was taken out from the furnace and cooled to room temperature using desiccators. The ignited sample powders were weighed again to determine the weight of calcinated impurities such as H_2O , H_2O^+ and CO_2 . Into 1 gram of each of the stored ignited sample powder was added 5 times of flux (x-ray flux-type) 66%:34% (66% lithium tetraborate: 34% lithium metaborate) to lower the vitrification temperature. The mixture was weighed and properly mixed in a Platinum dish and then placed in pre-set furnace (Eggon-2 automatic fuse bead maker) at $1,500^{\circ}\text{C}$ for 10 minutes to convert into a glass bead. Each glass bead produced was labeled and placed into the computerized XRF (Epsilon-5 pAnalytical model) to analyze the concentrations of the major oxides of elements in weight percentages. The X-ray fluorescence (XRF) spectrometer analysis was carried out in line with published standard procedure [15].

3. Result

The major oxide geochemical result is given in Table 1 and the bivariate plot of $\text{Log}_{10}(\text{Fe}_2\text{O}_3/\text{K}_2\text{O})$ on y-axis against $\text{Log}_{10}(\text{Si}_2\text{O}/\text{Al}_2\text{O}_3)$ on x-axis is shown on Fig. 2.

4. Discussion

In Table 1 the major oxides concentrations are as follows: SiO₂ (50.75 wt. % - 68.12 wt. %, average 57.26 wt. %), Al₂O₃ (10.36 wt. % - 15.96 wt. %, average 13.50 wt. %), Fe₂O₃ (3.96 wt. % - 14.20 wt. %, average 10.70 wt. %), TiO₂ (0 - 0.7 wt. %, average 0.26 wt. %), K₂O (4.45 wt. % - 8.85 wt. %, average 6.30 wt. %), CaO (1.08 wt. % - 8.70 wt. %, average 4.99 wt. %), MgO (0 wt. %), MnO (0.03 wt. % - 0.11 wt. %, average 0.07 wt. %), P₂O₅ (0.54 wt. % - 0.93 wt. %, average 0.73 wt. %). The calculated values of Log₁₀(Fe₂O₃/K₂O) are -0.18 to 0.39, average 0.22, while Log₁₀(Si₂O/Al₂O₃) values are 0.52 to 0.82, average 0.63. The plot of log₁₀(Si₂O/Al₂O₃) versus log₁₀(Fe₂O₃/K₂O) indicated that the sediments from the study area are 89.0% greywackes, 10.0% litharenites and 10.0% arkose. According to Amielwan and Lucas [18],

SiO₂/Al₂O₃ ratio ~3.0 represents basic igneous rock;

SiO₂/Al₂O₃ ratio: ~5.0 represents acid igneous rock;

SiO₂/Al₂O₃ ratio: >5.0 means the sediments progressively matured.

In Table 1 and Fig. 2, SiO₂/Al₂O₃ ratios range from 3.31 to 6.54 with average of 4.33 and that indicated that the stream sediments came from mixed sources of basic to acid igneous rocks. Lastly, Fe₂O₃/K₂O ratios range from 0.66 to 2.43 with average of 1.73 indicative that the sediments are mineralogically stable.

5. Conclusion

The oldest basement rocks in the study area are the Precambrian migmatites, biotite gneisses, granite gneisses and the Pan-African charnockites and biotite granites. The average compositions of major oxides are as follows: SiO₂ (57.26 wt. %), Al₂O₃ (13.50 wt. %), Fe₂O₃ (10.70 wt. %), TiO₂ (0.26 wt. %), K₂O (6.30 wt. %), CaO (4.99 wt. %), MgO (0 wt. %), MnO (0.07 wt. %), P₂O₅ (0.73 wt. %). The average values of SiO₂ suggest the provenance of the stream sediments to be from basic igneous or metamorphic rocks. The plot of Log₁₀(Fe₂O₃/K₂O) versus Log₁₀(Si₂O/Al₂O₃) indicates that the sediments from the study area are 89.0% greywackes, 10.0% litharenites and 10.0% arkose. Moreover, SiO₂/Al₂O₃ ratios range of 3.31 to 6.54 with average of 4.33 indicates that the stream sediments came from mixed sources of basic to acid igneous rocks. The stream sediments indicated progressive maturity, although majority of the sediments are not matured. Lastly, the average value of Fe₂O₃/K₂O (1.73) indicates that the sediments are mineralogically stable.

Table 1: Major Oxide Geochemical Result (Values in Wt.%)

Oxide	SL1	SL2	SL3	SL4	SL5	SL6	SL7	SL8	SL9	SL10	SL11	SL12	SL13	SL14	SL15	SL16	SL17	MIN.	MAX	MEAN
SiO ₂	60.36	67.76	61.92	68.12	57.83	60.85	56.37	59.86	53.62	56.79	51.13	55.47	53.91	50.75	52.84	51.08	54.78	50.75	68.12	57.26
Al ₂ O ₃	12.09	10.36	14.96	11.96	12.67	11.07	12.53	15.47	13	15.09	13.89	13.28	15.83	14.96	15.96	12.52	13.98	10.36	15.96	13.50
Fe ₂ O ₃	11.94	6.75	9.15	3.96	8.43	7.74	12.05	14.2	9.71	11.35	13.35	12.35	11.55	12.17	12.1	13.73	11.39	3.96	14.2	10.70
TiO ₂	0.28	0	0.06	0	0.34	0.7	0.19	0.35	0	0.19	0.27	0.41	0.18	0.45	0.32	0.63	0	0	0.7	0.26
K ₂ O	5.98	4.45	5.15	5.96	7.48	7.03	6.98	6.29	7.55	4.67	6.65	5.73	6.31	6.53	8.85	5.82	5.74	4.45	8.85	6.30
CaO	1.22	1.08	2.95	3.34	5.72	4.79	2.71	6.42	7.75	5.74	5.29	5.86	5.46	6.27	5.02	8.7	6.46	1.08	8.7	4.99
MgO	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00
MnO	0.05	0.05	0.05	0.03	0.05	0.03	0.06	0.09	0.07	0.08	0.09	0.08	0.07	0.06	0.09	0.11	0.09	0.03	0.11	0.07
P ₂ O ₅	0.75	0.85	0.54	0.62	0.73	0.79	0.93	0.69	0.85	0.59	0.8	0.6	0.66	0.8	0.77	0.76	0.72	0.54	0.93	0.73
Fe ₂ O ₃ /K ₂ O	2.00	1.52	1.78	0.66	1.13	1.10	1.73	2.26	1.29	2.43	2.01	2.16	1.83	1.86	1.37	2.36	1.98	0.66	2.43	1.73
Si ₂ O/Al ₂ O ₃	4.99	6.54	4.16	5.70	4.56	5.50	4.50	3.87	4.12	3.76	3.68	4.18	3.41	3.39	3.31	4.08	3.92	3.31	6.54	4.33
Log ₁₀ (Fe ₂ O ₃ /K ₂ O)	0.30	0.18	0.25	0.18	0.05	0.04	0.24	0.35	0.11	0.39	0.30	0.33	0.26	0.27	0.14	0.37	0.30	-0.18	0.39	0.22
Log ₁₀ (Si ₂ O/Al ₂ O ₃)	0.70	0.82	0.62	0.76	0.66	0.74	0.65	0.59	0.62	0.58	0.57	0.62	0.53	0.53	0.52	0.61	0.59	0.52	0.82	0.63

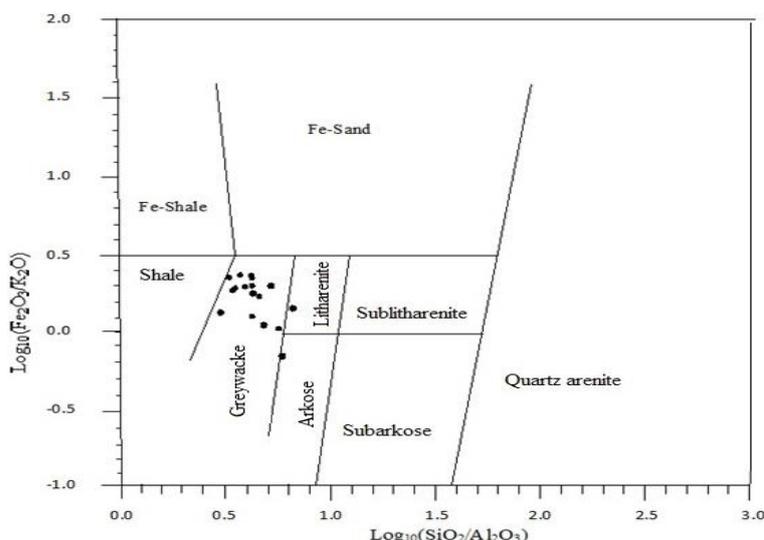


Fig. 2: Plot of log (SiO₂/Al₂O₃) Versus Log (Fe₂O₃/K₂O) (After [16-17]).

6. Conflicts of interest

The authors declare that there is no any conflict of interest in the execution of the research and its publication.

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