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To unite nuclear and sub-nuclear strong interactions

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Abstract

With reference to 'reciprocal' of the strong coupling constant and 'reduced Compton's wavelength' of the nucleon, we make an attempt to understand the background of nuclear charge radius, binding energy and stability.

Keywords: Strong Coupling Constant; Reduced Compton Wavelength of Nucleon; Beta Stability Line; Semi-Empirical Mass Formula (SEMF).

1. Introduction

The modern theory of strong interaction is quantum chromo dynamics (QCD) [1]. It explores baryons and mesons in broad view with 6 quarks and 8 gluons. According to QCD, the four important properties of strong interaction are: 1) color charge; 2) confinement; 3) asymptotic freedom [2]; 4) short-range nature ($<10^{-15}$ m). Color charge is assumed to be responsible for the strong force to act on quarks via the force carrying agent, gluon. Experimentally it is well established that, strength of strong force depends on the energy through the interaction or the distance between particles. At lower energies or longer distances: a) color charge strength increase; b) strong force becomes 'stronger'; c) nucleons can be considered as fundamental nuclear particles and quarks seem to be strongly bound within the nucleons leading to 'Quark confinement'. At high energies or short distances: a) color charge strength decreases; b) strong force gets 'weaker';3) colliding protons generate 'scattered free quarks leading to 'Quark Asymptotic freedom'. Based on these points, low energy nuclear scientists assume 'strong interaction' as a strange nuclear interaction associated with binding of nucleons and implications and its implications were not considered. High-energy nuclear scientists consider nucleons as composite states of quarks and try to understand the nature and strength of strong interaction at sub nuclear level.

At this juncture, one important question to be answered and reviewed at the basic level is: How to understand nuclear interactions in terms of sub nuclear interactions? Unfortunately, the famous nuclear models like, Liquid drop model and Fermi's gas model [3-6] are lagging in answering this question. To find a way, we would like to suggest that, by implementing the 'strong coupling constant' ($\alpha_r \approx 0.1186$), in low energy nuclear physics, nuclear binding energy and stability can be understood. In this new direction, we have developed interesting concepts and produced many semi empirical relations [7-9]. Even though it is in its budding stage, our model seems to be simple and realistic compared to the new integrated model proposed by N. Ghahramany et al [10-12]. It needs further study at a fundamental level.

2. Role of the strong coupling constant in low energy nuclear physics

We propose the following four assumptions.

- Nuclear binding energy can be understood with a single energy coefficient associated with 'reciprocal' of the strong coupling constant.
- 2) Characteristic nuclear radius can be expressed as, $\begin{pmatrix} 1 \\ 1 \end{pmatrix} \begin{pmatrix} b \\ b \end{pmatrix} \begin{pmatrix} 1 \\ 2b \end{pmatrix}$

$$R_{0} \cong \left(\frac{1}{\sqrt{\alpha_{s}}}\right) \left\{ \frac{h}{m_{p}c} + \frac{h}{m_{n}c} \right\} \cong \left(\frac{1}{\sqrt{\alpha_{s}}}\right) \left(\frac{2h}{m_{p}c}\right) \cong 1.22 \text{ fm.}$$

3) Characteristic nuclear binding energy potential can be,

$$B_{_{0}} \cong \left(\frac{1}{\alpha_{_{s}}}\right) \frac{e^{^{2}}}{4\pi\varepsilon_{_{0}}R_{_{0}}} \cong \left(\frac{\alpha}{2\sqrt{\alpha_{_{s}}}}\right) m_{_{p}}c^{^{2}} \cong 10.0 \text{ MeV where}$$

$$R_{_{0}} \approx 1.22 \text{ fm.}$$

4) (α_s) and $\exp\left(\frac{m_n - m_p}{m_e}\right) \cong 4\pi$ Seem to play a crucial role in

understanding beta stability line and nuclear binding energy.

3. About the semi empirical mass formula

Let A be the total number of nucleons, Z the number of protons and N the number of neutrons. According to the semi-empirical mass formula (SEMF), nuclear binding energy:

$$B = a_{r}A - a_{a}A^{23} - a_{c}\frac{Z(Z-1)}{A^{1/3}} - a_{a}\frac{(A-2Z)^{2}}{A} \pm \frac{a_{p}}{\sqrt{A}}$$
(1)

Here $a_r =$ volume energy coefficient, a_r is the surface energy coefficient, a_c is the coulomb energy coefficient, a_a is the asymmetry energy coefficient and a_r is the pairing energy coefficient. If we consider the sum of the volume energy, surface energy, coulomb energy, asymmetry energy and pairing energy, then the picture of a nucleus as a drop of incompressible liquid roughly accounts for the observed variation of binding energy of the nucleus.



By maximizing B(A,Z) with respect to Z, one can find the number of protons Z of the stable nucleus of atomic weight A as,

$$Z \approx \frac{A}{2 + (a_c/2a_a)A^{2/3}} \text{ and } A - 2Z \approx \frac{0.4A^2}{A + 200}$$
 (2)

By substituting the above value of Z back into B one obtains the binding energy as a function of the atomic weight, B(A). Maximizing B(A)/A with respect to A gives the nucleus which is most strongly bound or most stable.

4. Beta stability line with respect to strong coupling constant

If $\alpha_s \cong 0.1186$, for Z >8, close to the line of beta stability, for Z >8,

$$A_{s} \cong \left(Z + \sqrt{\frac{1}{\alpha_{s}}}\right)^{q/s} \cong \left(Z + 2.904\right)^{1/2}$$
(3)

for Z >16,

$$(A_{,} - 2Z) \cong (Z \beta + 1)^{2} - 4$$
where $\beta = \left(\frac{3}{5}\right)\alpha_{,}$.
$$(4$$

5. Beta stability line with respect to nucleon mass difference

With reference to nucleon and electron rest masses, we noticed that,

$$\exp\left(\frac{\left(m_{_{n}}-m_{_{p}}\right)c^{2}}{m_{_{e}}c^{2}}\right) \cong 12.5659102 \cong 4\pi$$
(5)

 $\begin{cases} \text{where, } m_{\mu}c^{2} \cong 939.565413 \text{ MeV,} \\ m_{\mu}c^{2} \cong 938.272081 \text{ MeV;} \quad m_{\mu}c^{2} \cong 0.5109989461 \text{ MeV} \end{cases}$

Based on this observation, beta stability line can be understood with the following empirical relations.

Let,
$$k \cong (1/4\pi)^2 \cong 0.006333$$
 (6)

For
$$Z \ge 2$$
,
 $A_{s} \cong 2Z + (1/4\pi)^{2} \cong 2Z + 0.006333Z^{2} \cong 2Z + kZ^{2}$
 $N_{s} \cong Z + (1/4\pi)^{2} \cong Z + 0.006333Z^{2} \cong Z + kZ^{2}$
 $A_{s} - 2Z \cong (1/4\pi)^{2} \cong kZ^{2}$ and $\frac{Z}{\sqrt{A_{s} - 2Z}} \cong 4\pi$

$$(7)$$

Based on these relations,

A)
$$\frac{(A_{s}-2Z)^{2}}{A_{s}} \approx k^{2}A_{s}(A_{s}-Z)\sqrt{Z}$$

B)
$$\frac{A_{s}^{1/2}(A_{s}-Z)^{1/4}Z^{1/8}}{\sqrt{A_{s}-2Z}} \approx \frac{1}{\sqrt{k}} \approx 4\pi$$
(8)

6. Relations for understanding nuclear binding energy

Based on the new integrated model proposed by N. Ghahramany et al [10-12] and with reference to relation (7), it is possible to show that, for $Z \cong (40 \text{ to } 83)$, close to the beta stability line,

$$\left[\frac{N_{s}^{2}-Z^{2}}{Z}\right] \cong kZA_{s}$$
(9)

Based on this strange and simple relation and with reference to our recent publications [8], [9], close to the beta stability line, numerically it is possible to show that, for $(Z \ge 24)$, if $\alpha_s \approx 0.1186$ and $R_0 \approx 1.22$ fm,

$$(B)_{A} \approx \left[A - A^{V^{3}} - \frac{kA\sqrt{NZ}}{3.4} \right] \times 10.0 \text{ MeV}$$

$$\approx \left\{ A - A^{V^{3}} - \left[(\alpha_{s}) \left(\frac{a_{c}}{2a_{s}} \right) (A\sqrt{NZ}) \right] \right\} \times 10.0 \text{ MeV}$$

$$(10)$$

where,
$$\alpha_{s}\left(\frac{a_{c}}{2a_{a}}\right) \cong \frac{k}{3.4}$$
. It is for further study

For $(Z \ge 24)$, binding energy per nucleon can be expressed as:

$$\frac{(B)_{\lambda}}{A} \simeq \left[1 - A^{-2/3} - \frac{k\sqrt{NZ}}{3.4}\right] \times 10.0 \text{ MeV}$$
(11)

a) See table 1 for nuclear binding energy of stable and unstable isotopes of Z=23 to 100 estimated from relations (7) and (10).

b) See table 2 for nuclear binding energy of natural isotopes of Z=30, 40, 50, 60, 70, 80 and 92 estimated from relation (10).

 c) See table 3 for nuclear binding energy of natural isotopes of Z=25 to 83 estimated from relation (10).

Table 1: Nuclear Binding Energy of $Z = 23$ To 100 Estimated From Relations	(7)) and ((10))
Lable 1. Nuclear Difficing Energy of $L = 25$ 10 100 Estimated 110 m Kelations	U) and (, I U	,

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Proton	Mass	Neutron	Estimated	Actual[10] or reference [4]	Error
number	number	number	binding energy(MeV)	binding energy (MeV)	(MeV)
23	49	26	431.1	426.34	-4.7
24	52	28	457.6	456.349	-1.2
25	54	29	475.1	472.33	-2.8
26	56	30	492.6	492.258	-0.4
27	59	32	518.8	517.313	-1.5
28	61	33	536.1	534.666	-1.4
29	63	34	553.4	551.385	-2.0
30	66	36	579.2	578.136	-1.1
31	68	37	596.3	590.61	-5.7
32	70	38	613.3	610.521	-2.8
33	73	40	638.8	634.34	-4.5
34	75	41	655.7	651.02	-4.7
35	78	43	680.9	676.11	-4.8
36	80	44	697.6	695.434	-2.2

27	92	16	702.6	720.46	2.1
37	85	40	722.0	720.40	-2.1
38	85	47	739.1	737.85	-1.3
20	00	40	762.0	762.99	0.0
39	88	49	103.9	705.88	0.0
40	90	50	780.2	783.893	3.7
41	02	50	804 7	205 765	1.1
41	93	32	004.7	803.703	1.1
42	95	53	820.9	821.625	0.7
42	09	<i>E E</i>	945 1	944 4	0.7
43	98	55	845.1	844.4	-0.7
44	100	56	861.1	861 928	0.8
15	100	50	007.1	001.520	0.0
45	103	58	885.1	884.163	-0.9
46	105	50	900.9	800.01/	-1.0
40	105	57	000.0	0)).)14	-1.0
47	108	61	924.7	922.2	-2.5
18	111	63	0/8 2	047 622	0.6
40	111	05	940.2	947.022	-0.0
49	113	64	963.8	963.094	-0.7
50	116	66	097.1	000 601	16
30	110	00	987.1	900.004	1.0
51	118	67	1002.5	1000.48	-2.0
50	101	60	1025 5	1024.42	1.1
32	121	09	1025.5	1024.45	-1.1
53	124	71	1048.5	1046.32	-2.1
51	120	70	10(2.5	10/2 000	0.4
54	120	12	1005.5	1065.909	0.4
55	129	74	1086.2	1085.08	-1.1
50	120	76	1100 7	1110.020	1.4
56	132	/6	1108.7	1110.038	1.4
57	135	78	1131.0	1131.00	0.0
50	100	70	1145 7	1145.7	0.0
58	137	/9	1145.7	1145./	0.0
50	140	81	1167.8	1168.67	0.0
57	140	01	1107.0	1100.07	0.5
60	143	83	1189.7	1191.266	1.5
61	146	85	1211.5	1200 52	2.0
01	140	85	1211.5	1209.32	-2.0
62	148	86	1225.8	1225.392	-0.4
62	151	00	1247.2	1244 141	2.2
03	131	00	1247.5	1244.141	-3.2
64	154	90	1268.7	1266.627	-2.1
65	157	02	1290.0	1007 20	25
03	137	92	1289.9	1267.36	-2.3
66	160	94	1311.0	1309.455	-1.5
(7	1(2)	05	12247	1201 10	20
07	162	95	1324.7	1521.18	-3.0
68	165	97	1345.5	1343.08	-2.5
60	1.00	00	12(6.2	12(2.21	2.0
69	168	99	1366.2	1363.31	-2.9
70	171	101	1386.7	1384 744	-19
70	171	102	1407.0	1404.44	1.5
71	174	103	1407.0	1404.44	-2.6
72	177	105	1427.2	1425 185	-2.0
72	177	105	1+27.2	1425.105	-2.0
73	180	107	1447.2	1444.663	-2.6
74	183	109	1467.1	1465 525	-16
74	105	105	1407.1	1403.323	-1.0
75	186	111	1486.8	1484.63	-2.2
76	180	113	1506 /	1505 007	1 /
70	189	115	1500.4	1505.007	-1.4
77	192	115	1525.8	1523.81	-2.0
78	105	117	1545.0	1545 682	07
78	195	117	1343.0	1545.062	0.7
79	198	119	1564.1	1564.94	0.8
80	201	121	1583 1	1587 /11	13
80	201	121	1565.1	1307.411	4.5
81	204	123	1601.9	1606.87	5.0
82	207	125	1620.5	1629.063	86
02	207	123	1020.3	1027.003	0.0
83	210	127	1639.0	1643.94	5.0
84	213	120	1657.3	1650 72	2.4
04	215	129	1057.5	1057.12	2.4
85	216	131	1675.4	1673.42	-2.0
86	210	133	1603 5	1600 50	2.0
00	219	155	1093.5	1070.37	-2.9
87	222	135	1711.3	1706.49	-4.8
88	225	137	1729.0	1724 18	18
00	223	137	1729.0	1/24.10	-+.0
89	228	139	1746.6	1740.67	-5.9
00	231	141	1763.0	1750 14	18
90	231	141	1703.9	1757.14	-+.0
91	234	143	1781.2	1776.08	-5.1
02	220	146	1904.2	1801.60	26
92	230	140	1004.2	1001.09	-2.0
93	241	148	1821.1	1817.31	-3.8
04	211	150	1027.0	1025 45	2.4
94	244	150	1837.8	1833.43	-2.4
95	247	152	1854.4	1851.73	-2.7
06	250	154	1070.0	10(0.07	1.0
90	250	154	18/0.8	1808.97	-1.8
97	254	157	1892.8	1888.79	-4.0
00	257	150	1000.0	1006.10	0.7
98	257	159	1908.9	1906.19	-2.7
99	260	161	1924.8	1922.2	-26
	200	101	1727.0	1722.2	2.0
100	263	163	1940.5	1939.52	-1.0

Table 2: Nuclear Binding Energy of Natural Isotopes of Z = 30, 40, 50, 60, 70, 80, 92 Estimated from Relation (10)					
Proton	Mass	Neutron	Estimated binding energy	Actual [10] binding energy	Error
number	number	number	(MeV)	(MeV)	(MeV)
30	64	34	561.9	559.098	-2.83
30	66	36	579.2	578.136	-1.05
30	67	37	587.8	585.189	-2.62
30	68	38	596.4	595.387	-1.03
30	70	40	613.6	611.087	-2.53
40	90	50	780.2	783.893	3.68
40	91	51	788.5	791.087	2.62
40	92	52	796.7	799.722	3.02
40	94	54	813.2	814.677	1.52
40	96	56	829.6	828.996	-0.59
50	112	62	955.6	953.532	-2.11
50	114	64	971.4	971.574	0.18

238

146

92

50	115	65	979.3	979.121	-0.13
50	116	66	987.1	988.684	1.58
50	117	67	995.0	995.627	0.67
50	118	68	1002.8	1004.955	2.16
50	119	69	1010.6	1011.438	0.82
50	120	70	1018.4	1020.546	2.11
50	122	72	1034.1	1035.53	1.47
50	124	74	1049.6	1049.963	0.32
60	142	82	1182.3	1185.142	2.84
60	143	83	1189.7	1191.266	1.53
60	144	84	1197.2	1199.083	1.92
60	145	85	1204.6	1204.838	0.25
60	146	86	1212.0	1212.403	0.41
60	148	88	1226.8	1225.028	-1.76
60	150	90	1241.6	1237.448	-4.11
70	168	98	1365.6	1362.793	-2.85
70	170	100	1379.7	1378.13	-1.55
70	171	101	1386.7	1384.744	-1.94
70	172	102	1393.7	1392.764	-0.91
70	173	103	1400.7	1399.131	-1.53
70	174	104	1407.6	1406.595	-1.05
70	176	106	1421.6	1419.283	-2.29
80	196	116	1550.2	1551.218	1.00
80	198	118	1563.4	1566.489	3.10
80	199	119	1570.0	1573.153	3.20
80	200	120	1576.5	1581.181	4.66
80	201	121	1583.1	1587.411	4.34
80	202	122	1589.6	1595.165	5.55
80	204	124	1602.7	1608.652	5.98
92	234	142	1780.2	1778.567	-1.63
92	235	143	1786.2	1783.864	-2.36

	Table 3	: Nuclear Binding En	ergy of Natural Isotopes of Z = 25 To83	Estimated from Relation (10)	
Proton	Mass	Neutron	Estimated binding energy	Actual [10] binding energy	Error
number	number	number	(MeV)	(MeV)	(MeV)
25	55	30	483.9	482.075	-1.84
27	59	32	518.8	517.313	-1.45
29	65	36	570.7	569.212	-1.46
31	69	38	604.9	601.996	-2.88
33	75	42	655.8	652.564	-3.26
35	79	44	689.3	686.321	-3.03
37	87	50	756.0	757.856	1.87
39	89	50	772.1	775.538	3.39
41	93	52	804.7	805.765	1.06
45	103	58	885.1	884.163	-0.95
47	109	62	932.6	931.727	-0.91
49	115	66	979.6	979.404	-0.15
51	123	72	1041.4	1042.097	0.66
53	127	74	1071.6	1072.577	0.99
55	133	78	1116.7	1118.528	1.83
57	139	82	1161.2	1164.551	3.36
59	141	82	1175.3	1177.919	2.64
63	153	90	1261.9	1258.998	-2.93
65	159	94	1304.3	1302.027	-2.30
67	165	98	1346.1	1344.256	-1.86
69	169	100	1373.2	1371.352	-1.88
71	175	104	1414.0	1412.106	-1.86
73	181	108	1454.1	1452.24	-1.84
75	187	112	1493.6	1491.877	-1.70
77	193	116	1532.5	1532.058	-0.40
79	197	118	1557.5	1559.386	1.86
81	203	122	1595.3	1600.87	5.52
83	209	126	1632.5	1640.23	7.68

1804.2

With an error bar of 5 MeV, to some extent, relation (10) can be applied to light atomic nuclides and for Z = 3 to 100, modified relation can be expressed as,

$$(B)_{A} \cong \left[A - A^{\frac{1}{3}} - \frac{kA\sqrt{NZ}}{3.35} - 1\right] \times 10.09 \text{ MeV}$$
 (12)

Estimated data can be compared with first four terms of the semi empirical mass formula.

7. Discussion

a) By considering coulombic repulsions and with reference to our earlier publications [7], [8], close to the beta stability line, binding energy can also be expressed with: for $(Z \ge 24)$,

1801.69

$$(B)_{A} \cong \left[A - kN\left(\frac{kAA_{A}}{\sqrt{Z}} + 1\right)\right] \times 8.9 \text{ MeV}$$
 (13)

-2.56

 $R_0 \cong 1.22 \text{ fm}$ and

where
$$\left\{ \left\{ \left[\left(\frac{1}{\alpha_{s}}\right) \frac{e^{2}}{4\pi\varepsilon_{0}R_{0}} \right] - \left[\frac{e^{2}}{4\pi\varepsilon_{0}R_{0}}\right] \right\} \cong 8.9 \text{ MeV} \right\}$$

- b) By fine tuning the values of $(\alpha_{s} \text{ and } R_{o})$ and by considering even-odd corrections and shell corrections, accuracy can be improved.
- c) With a suitable mathematical relation and with a single energy coefficient lying in between (8.9 to 10) MeV, it is possible to estimate nuclear binding energy.
- d) Close to the beta stability line, ignoring the pairing energy term, semi empirical mass formula energy coefficients can

be expressed as: $a_{y} \approx a_{z} \approx a_{a} \approx \frac{3}{2} \times 10.0 \text{ MeV} \approx 15.0 \text{ MeV}$ and

 $a_c \approx 0.73$ MeV.

8. Conclusion

Understanding and estimating nuclear binding energy and stability with 'sub-nuclear strong interaction' seems to be quite interesting and needs a serious consideration at a fundamental level. We believe that, results obtained from above relations are simple to understand and seem to be more physical and relatively closer to the experimental data. With further research, current nuclear models and strong interaction concepts, pertaining to high-energy physics can be studied in a unified manner, and a realistic nuclear model can be developed.

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